

$$(P5.1) \text{ COP} = \text{coef. of performance} = \frac{\dot{Q}_C}{\dot{W}_{S,net}}$$

Using state numbers of Fig 5.8-5.9. P-H plot will look like Fig 5.9. Use P-H chart and table from Appendix E. Saturated values are from the table:

state 2 is satV at $-40^\circ\text{C} \rightarrow H_2 = 372 \text{ kJ/kg}$ (chart)

state 3, outlet of the reversible compressor is found by following the isentropic line to 40°C , where $H_3' = 438 \text{ kJ/kg}$.

state 4 is satL at $H_4 = 256 \text{ kJ/kg}$ (table)

state 1, $H_1 = H_4$

$$Q_C = (H_2 - H_1) = 372 - 256 = 116 \text{ kJ/kg}$$

$$W_S = (H_3 - H_2) = 438 - 372 = 66 \text{ kJ/kg}$$

$$\Rightarrow COP = \frac{\dot{Q}_C}{\dot{W}_S} = \frac{116}{66} = 1.76$$

$$Q_C = COP (W_S) = 1.76 (9000 \text{ J/day}) = 16 \text{ kJ/day}$$

(P5.2) Methane, See Fig 5.8-5.9. Methane chart Appendix E.

$$T_{evap} = -280^\circ\text{F}, P = 0.032 \text{ MPa}$$

$$\Rightarrow H_2^{satV} = 339 \text{ Btu/lbm}$$

$$S_2^{satV} = 2.35 \text{ Btu/lb}^\circ\text{F}$$

Pressure of 3' isn't given, however, it will be the same as the condenser outlet (P_4) since the condenser will be considered to be isobaric. State 4 will be satL at 40°F , so $P_4 = P_3' = 2 \text{ MPa}$. Follow isentropic line from the compressor inlet state to 2 MPa . At $P_3' = 2 \text{ MPa}$, and $S_3' = 2.35 \text{ Btu/lbm}^\circ\text{F}$, $H_3' = 485 \text{ Btu/lbm}$, and $T_3' = 40^\circ\text{F}$. At the outlet of the condenser, the fluid will be saturated liquid at 2 MPa ,

$$T_4 = -160^\circ\text{F}, P_4 = 2 \text{ MPa}$$

$$\Rightarrow H_4^{satL} = 218 \text{ Btu/lbm}$$

$$S_4^{satL} = 1.54 \text{ Btu/lb}^\circ\text{F}$$

State	T($^\circ\text{F}$)	P(MPa)	H(Btu/lb)	S(Btu/lb $^\circ\text{F}$)
1	-280	0.032	218	
2 Sat'd vap	-280	0.032	339	2.35
3'	40	2	485	2.35
4	-160	2	218	1.54

$$COP = \frac{H_2 - H_1}{H_3' - H_2} = \frac{Q_C}{W_{S,net}} = \frac{H_2 - H_4}{H_3' - H_2} = \frac{339 - 218}{485 - 339}$$

$$\Rightarrow COP = 0.83$$

(P5.3) Rankine Cycle, Fig 5.2

temperatures 200°C & 99.6°C , Sat vapor from Turbine outlet can be found in the Sat Pressure table Appendix E.

Chapter 5 Practice Problems

The turbine is assumed to be adiabatic and reversible since no other specifications are given. The inlet state will then have the same entropy as the outlet. Since the boiler/superheater outlet is to operate at $T_3 = 200\text{ C}$, and $S_3 = 7.3592$, we must hunt in the superheated tables to find this combination. This will occur between 0.2 and 0.3 MPa. Interpolating, the value of H can be found. Values are tabulated below:

State	T(°C)	P(MPa)	H(kJ/kg)	S(kJ/kg·K)	V(m ³ /kg)
4' sat vap	99.6	0.098	2675	7.359	
5 sat liq.	99.6	0.098	417	1.3028	0.001043
	3	200	2867	7.3592	
	6		417.18		

$$W'_{S,turbine} = (H'_4 - H_3) = 2675 - 2867 = -192\text{kJ} / \text{kg}$$

$$\Delta H_{pump} = W_{S,pump} = \int VdP = V\Delta P$$

$$\Rightarrow \Delta H_{pump} = \frac{1043\text{cm}^3 * (0.271 - 0.098)}{\text{kg}} * \frac{1\text{J}}{\text{cm}^3 * \text{MPa}} * \frac{\text{kJ}}{10^3\text{J}}$$

$$\Rightarrow \Delta H_{pump} = 0.180\text{kJ} / \text{kg}.$$

$$\Rightarrow W'_{S,net} = -192 + 0.180 = -191.82\text{kJ} / \text{kg}$$

$$\& \eta = \frac{-W'_{S,net}}{Q_{boiler}} = \frac{-(-191.82)}{(H_3 - H_6)} = \frac{-(-191.82)}{(2867 - H_6)}$$

$$\Rightarrow H_6 = 417 + 0.18 = 417.18\text{kJ} / \text{kg}$$

$$\Rightarrow \eta = \frac{-(-191.82)}{(2867 - 417.18)} = 0.0783$$

$$\Rightarrow \eta = 7.83\%$$

(P5.6) Inlet: 300 psia/14.504 = 20.68 bar; (550 °F + 460)/1.8-273.15 = 288 °C = 561.15K

Steam.xls => H1=2993.9; S1 = 6.7004

Outlet: S2 = 1.3072 + 0.95(7.3541 - 1.3072) = 7.053; H2 = 419.17+0.95*2256.4=2562.8

W = 2562.8-2993.9 = -431.1

S2' = S1 => q' = (6.7004-1.3072)/(7.3541-1.3072) = 0.8919

H2' = 419.17+0.8919*2256.4=2431.7 => W' = 2431.7-2993.9 = -562.2

=> $\eta_E = 431.1/562.2 = 76.7\%$

(b) Base Rankine cycle on actual turbine => $W_p = 1.9\text{J/g}$; $Q_H = 2993.9 - (419.17 + 1.9) = 2572.8$

$\eta_\theta = (431.1 - 1.9)/2572.8 = 16.7\%$